

Stability Indicating Assay Method for the Quantitative Determination of Olaparib in Bulk and Pharmaceutical Dosage Form

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ABSTRACT I

Objectives: Olaparib is an orally active poly (ADP-ribose) PARP (polymerases) inhibitor known to destroy cancer cells with BRCA1 or BRCA2 deficiency. An authentic, fast, distinct, and reliable reverse phase-high performance liquid chromatography (RP-HPLC) method was developed and promptly validated in tablet formulations for olaparib estimation.

Materials and Methods: The proposed method focuses on the separation of olaparib in reverse phase mode using a Waters symmetry C18 (150 x 4.6 mm, $5 \mu m$) analytical column with a flow rate of 1.0 mL/min and the injection volume was kept at 20 μ L. The optimized mobile phase consists of ammonium acetate buffer (pH adjusted to 3.5 by glacial acetic acid): methanol in the ratio of 50:50 v/v.

Results: The eluents were measured at 254 nm and the retention time for the drug encircled was about 4.32 min. The stress degradation studies of olaparib were conducted under acidic, alkaline, oxidative, photolytic and thermal conditions to demonstrate the stability of the drug. The regression value of 0.998 showed that the developed method was linear over the range of 80 μ g/mL to 120 μ g/mL. The developed RP-HPLC method is accurate and precise. The method was statistically validated as *per* International Conference on Harmonization guidelines.

Conclusion: The proposed method is suitable and can be applied for the quantitative estimation of olaparib without any interference of the excipients used in the drug formulations.

Key words: Olaparib, poly ADP-ribose polymerase (PARP) inhibitor, RP-HPLC, waters, ICH and validation

INTRODUCTION

During the last decade, inactivation of poly (ADP-ribose) polymerase (PARP), a nuclear enzyme associated with many operations including DNA repair and cell death, has emerged as a possible individualized cancer therapeutic approach.¹⁻⁴ In cancer cells with a defective DNA damage repair system, such as those produced by *BRCA* gene mutations, PARP inhibitors, a new class of anticancer drugs, can cause tumor-specific synthetic lethality.⁵⁻⁸ Olaparib (Figure 1), veliparib, niraparib, and rucaparib are potent PARP inhibitors that have recently moved through advanced clinical studies as combination and/ or solo-targeted therapies, especially in breast and ovarian malignancies. Olaparib (Lynparza®) was the first medication to be approved for use in individuals with BRCA-mutated ovarian

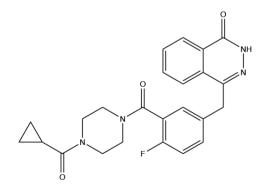


Figure 1. Chemical structure of olaparib

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cancer by the European Commission (2014) and the United States Food and Drug Administration (2015).^{5,9,10}

PARP inhibitors hold a lot of therapeutic potential and will likely be employed in many cancer therapies in the future.¹⁰ However, preclinical and clinical studies have revealed that tumor cell sensitivity to PARP inhibitors varies significantly, indicating that treatment efficacy must be enhanced.¹¹ Because PARP is an intracellular target, a crucial element influencing tumour cell sensitivity and the efficacy of a PARP-targeted treatment is the quantity of PARP inhibitors reaching the intracellular compartment.¹² PARP inhibitors, like any other intracellular target medicine, are affected by processes such as excretion, metabolism,¹³ drug absorption, and expression/upregulation of transmembrane drug efflux transporters.^{14,15} The latter, which is particularly significant for PARP inhibitors, was discovered as a key resistance mechanism during early preclinical trials.¹⁶⁻¹⁸

Analytical method validation assures that diverse high performance liquid chromatography (HPLC) analytical procedures provide consistent and reproducible results; it is an important stage in the development of novel dosage forms since it provides information on accuracy, linearity, precision, detection, and quantification limits. "The goal of validation of an analytical procedure is to demonstrate that it is suitable for its intended purpose", according with the International Conference on Harmonization (ICH) guideline. Validation data must now be sent to the appropriate authorities during the medication development process. The validation of analytical methods is governed by a set of guidelines from the ICH and the United States Pharmacopeia.

Olaparib has not yet been formally included in any of the official pharmacopeias, and there is no documented reverse phase (RP)-HPLC technique for quantifying olaparib in pharmaceutical formulations, according to a comprehensive literature assessment. However, only a few techniques for estimating olaparib concentrations in human plasma using UHPLC10 and liquid chromatography-tandem mass spectrometry (LC-MS/MS) have been reported. Lower This work aimed to create a genuine, quick, distinct, and reliable analytical technique for quantifying olaparib in pharmaceutical formulations using RP-HPLC, which was validated according to ICH guidelines. A proven method for quantifying olaparib in bulk and pharmaceutical formulations was successfully implemented.

MATERIALS AND METHODS

Chemical and reagents

The various laboratory batch samples and reference standard (99.92%) of olaparib were provided by AstraZeneca Pharma. HPLC grade methanol was procured from Merck Sigma-Aldrich. Milli-Q purified water a Milli-Q plus purification system from Millipore was utilized during experimental studies. HPLC grade ammonium acetate was obtained from Rankem.

Instrument

In the study, a Waters HPLC 2695 sequence with a pump, auto sampler, auto injector, variable wavelength detector, and 2690

PDA detector with thermostatic column compartment was used. Operation control of the instrument and data collection was done by empower 3 software.

Optimization of chromatographic conditions

The HPLC method was optimized with objective to estimation olaparib in tablet formulation. Several mobile phases in isocratic mode, along with various columns, were considered to achieve a sharp peak with the base line. The tailing factor, the sharpness of the peak, and symmetry were considered for selectivity, sensitivity, and appropriate chromatographic conditions suitable for the column and the mobile phase. Different flow rates were also attempted and fixed at 1 mL/min for the optimized method. The eluents were also checked for their maximum absorbance in the PDA detector and fixed at 254 nm as a detection wavelength. The temperature of the column was maintained at 25°C.

Preparation of mobile phase

The mobile phase was prepared at a ratio of 50:50 v/v of buffer and methanol. Ammonium acetate buffer was created by dissolving 1.0 g of ammonium acetate in a sufficient volume of Milli-Q water (1000 mL). The pH of the solution was adjusted to 3.5 using glacial acetic acid. The mobile phase was degassed by sonication and it filtered using a 0.45 μ membrane filter. Methanol was used as the diluent. The ultraviolet detector was set at a wavelength of 254 nm.

Preparation of standard solutions

The standard stock solution of olaparib was prepared by weighing 25 mg olaparib into a 25 mL volumetric flask, sonicating until dissolved, and finally, the volume was made up to 25 mL with methanol. Appropriate dilutions were made from the above 1.0 mg/mL solution and transferred to a 100 mL volumetric flask, where the final volume was made by the mobile phase. Before the filling the vials for chromatographic analysis, the solution was passed through a membrane filter of diameter 0.45 $\mu.$

Sample solution preparation for estimating marketed tablet formulation

Twenty tablets were accurately weighed, powdered and was added to 25 mL of diluent in a volumetric flask followed by sonication till it was completely dissolved. Finally, the volume was made up to 50 mL. Appropriate dilution was made to obtain a concentration of 100 μ g/mL as a stock solution. Different dilutions were prepared from the stock solution and chromatographic analysis was carried out. Before filling the vial for chromatographic analysis, the solution was filtered *via* a 0.45 μ membrane filter.

Method validation

The optimized method developed for olaparib was validated in accordance with the ICH guideline Q2 (R1) for evaluating linearity, precision, accuracy, specificity, robustness, ruggedness, system suitability, analytical solution stability, and force degradation.

Linearity

The linearity range of an analytical method was assessed by injecting the standard dilution in duplicates over five different concentrations made in the range of 80 μ g/mL to 120 μ g/mL. The calibration curve was plotted with the analyte peak area against the analyte concentration to ensure the linearity of the analytical method being developed.

Precision and intermediate precision

The intra-and inter-day precision was determined in terms of the peak area difference of drug solutions for three consecutive days. A relative standard deviation (RSD) was calculated from the alteration of peak area to represent the intra- and inter-day precision.

Intra- and inter-day precision were performed at three different concentration levels of 80, 100, and 120 μ g/mL. The repeatability study was performed by injecting six replicates of standard preparations of concentration 100 μ g/mL.

Accuracy

The accuracy of the developed method was verified by spiking olaparib, which was performed by spiking olaparib with a standard at three different concentrations: 90%, 100%, and 110%. Triplicate analysis of these samples was performed and the results were in the form of RSD% and recovery percentage.

Specificity

The specificity of the method was established by analyzing standard substances against potent interferences. Specificity was assessed by injecting standard, sample, placebo, and blank preparations into HPLC. The recovery was measured.

Robustness

Alterations were made deliberately in chromatographic parameters such as the composition of the mobile phase, mobile phase pH, and flow rate. These variations were evaluated for column efficiency, asymmetry factor along with their RSD%.

Ruggedness

Different analysts were considered for the ruggedness study. Solutions of 100 µg/mL were prepared and injected by two different analysts and the result was given in the form of RSD%.

Assay of standard formulation of lynparza (olaparib)

Twenty tablets were weighed and crushed to powder. A quantity of this powder equivalent to 50 mg was taken in a 50 mL volumetric flask to which 25 mL diluent was added. The solution was sonicated for 30 min, and the volume was adjusted up to the mark with diluent. The solution was further diluted to obtain a concentration of 100 μ g/mL of olaparib. Before the filling the vials for chromatographic analysis, the solution was passed through a 0.45 μ membrane filter.

System suitability

System suitability parameters such as tailing factor, resolution, theoretical plates, and percent RSD were assessed by injecting a blank observed by six replicates of the olaparib standard as well as sample solutions at a concentration of $100~\mu g/mL$.

Limit of detection (LOD) and limit of quantification (LOQ)

LOD and LOQ were calculated from the calibration according to the formulas mentioned;

LOD = 3.3 SD/Slope

LOQ = 10 SD/Slope

or detection limit= 3.3 σ /s, quantification limit= 10 σ /s, where σ is the standard deviation of y- intercept of the regression line, and s is the slope of the calibration curve.

Solution stability

The stability of the analytical solution was established by injecting the standard solution at a periodic interval of 48 h by maintaining the temperature of the auto sampler at room temperature. The solution response was measured and the percentage differences in the peak area have been calculated.

Force degradation study

According to the ICH guideline Q1A (R2), a force degradation study of olaparib was conducted under stress conditions. The olaparib active pharmaceutical ingredient (100 μ g/mL) was subjected to hydrolysis (acid and alkali), peroxide, thermal, and photolytic degradation for the stability study.

Acid degradation

A standard solution of 5 mL olaparib was taken in a 50 mL clean and dry volumetric flask. To the volumetric flask, 2.5 mL of 5 M HCl was added and kept for 30 min. After the completion of 30 min, neutralize the solution was with 2.5 mL of 5 M NaOH and the 50 mL volume was made by the mobile phase. Finally, filtering the solution was done with a 0.45 μm filter. The filtered solution was introduced into HPLC and the peak area was compared with the standard chromatogram.

Alkali degradation

A standard solution of 5 mL olaparib was taken in a 50 mL clean and dry volumetric flask. To the volumetric flask, 2.5 mL of 5 M NaOH was added and kept for 30 min. After 30 min, the solution was neutralized with 2.5 mL of 5 M HCl, and the mobile phase was responsible for 50 mL volume. Finally, the prepared solution was filtered with a 0.45 μm filter. The filtered solution was introduced into HPLC and the peak area was compared with the standard chromatogram.

Peroxide degradation

A standard solution of 1.0 mL olaparib was taken in a 10 mL clean and dry volumetric flask. To the volumetric flask, 1 mL of 30% $\rm H_2O_2$ was added and kept in the flask for 30 min. After the completion of the 30 min, a volume of 10 mL was made by the mobile phase. Finally, the prepared solution was filtered with a 0.45 μm filter. The filtered solution was then introduced into HPLC and the peak area was compared with the standard chromatogram.

Thermal degradation

The powdered sample of olaparib was spread on a petri dish with 1.0 mm thickness and kept at 70°C in a hot air oven for 3 h. 25 mg of the sample was taken in a 25 mL clean and dry

volumetric flask. 10 mL solution was pipette out and the volume of 100 mL was made by the mobile phase. Finally, the prepared solution was filtered with a 0.45 μ m filter. The filtered solution was introduced into HPLC and the peak area was compared with the standard chromatogram.

Photolytic degradation

The powdered sample of olaparib was spread on a petri dish 1.0 mm thickness and kept in direct sunlight for 3 h. 25 mg of the sample was taken in a 25 mL clean and dry volumetric flask. 10 mL solution was pipette out and the volume of 100 mL was made by the mobile phase. Finally, the prepared solution was filtered with a 0.45 μm filter. The filtered solution was introduced into HPLC and the peak area was compared with the standard chromatogram.

Statistical analysis

Statistical data were not used during the experiments.

RESULTS

Method development

Chromatographic separation

Various chromatographic systems (RP-HPLC) were considered to optimize the separation of olaparib. Olaparib separation was performed on the column C18 (150 mm x 4.6 mm, 5 μ m). The mobile phase was a combination that included 500 mL methanol, 500 mL buffer, and 1.0 mL of glacial acetic acid, pH adjusted to 3.5 \pm 0.05 with ammonium acetate. The flow rate was set to 1.0 mL/min and the detector was set to 254 nm. The injection volume was kept at 20 μ L. The retention time for olaparib was found to be 4.32 min as shown in Figure 2.

Calibration curve

The calibration curve was prepared and evaluated using the least square method within the Microsoft Excel® program. The coefficient of determination (R^2), slope and intercept for olaparib were 0.998, 23599, and 66731, respectively. The linear equation was found to be y = 23599x + 66731 and the calibration curve is shown in Figure 2.

Validation of the method

Linearity

The analytical calibration curve was plotted for olaparib and was found to be linear in the specified ranges (80-120 μ g/mL) indicating a correlation coefficient R² of 0.99 (acceptance limit λ 0.98). The slope of the straight line was found to be 23599 and the intercept was found to be 66731. The results are reported in Table 1 and the calibration curve is shown in Figure 2.

Precision and intermediate precision

The precision value was reported in terms of RSD%. The RSD% for olaparib was found to be $\langle 2.0\%$ for both inter- and intra-day precision, indicating satisfactory precision (Table 2). The inter-day precision was found to be within 0.05-0.98, whereas intra-day precision was found to be within 0.06-0.43.

Accuracy

The accuracy of this method is determined by a recovery study conducted using standard addition methods at six concentration levels, first 90%, 100%, and 110%. The spiked sample solutions were assayed in triplicate and the obtained results were compared with the expected results and expressed as the

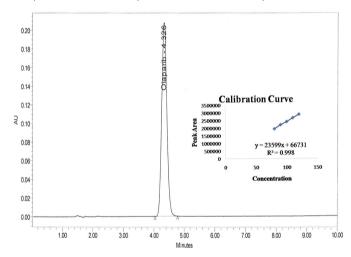


Figure 2. HPLC chromatogram of olaparib at 254 nm. Inset: Calibration curve of olaparib

HPLC: High performance liquid chromatography

Table 1. Linear regression equation generated from validation of olaparib: Slope, intercept, and coefficient of determination						
Concentration	Concentration	Peak	Peak area	Peak area	Average	
(mg/mL)	(µg/mL)	Area 1	2	3	Area	
0.080	80	1938729	1945764	1942428	1942307	
0.090	90	2215489	2215608	2210027	2213708	
0.100	100	2413316	2411985	2415423	2413575	
0.110	110	2671899	2672202	2663858	2669319.600	
0.120	120	2894206	2891855	2897362	2894474.300	
Slope	-	-	-	-	23599	
Intercept	-	-	-	-	66731	
R ² (correlation)	(Coefficient)	-	-	-	0.998	

percentage of recovery reported in Table 3. The recovery was found to be within the limit.

Specificity

The developed analytical method should reflect that there was no interference due to the presence of excipients in the formulation. The recovery and RSD% of olaparib were measured and were within the limits summarized in Table 4. The recovery was found to be 100.81-101.71 and RSD% was within 0.10-0.80.

Robustness

The method was found to be robust, ensuring that upon applying small variations to the chromatographic conditions in terms of flow rate, mobile phase composition, and pH variation in the mobile phase, no significant changes are detected. The robustness data were expressed in terms of RSD% was found to be 0.68 and is given in Table 5.

Ruggedness

Analyst 1 and analyst 2 performed the ruggedness test and the results are summarized in Table 6 as RSD% was found to be 1.48.

Table 2a. Intra-day precision						
Concentration (µg/mL)	Peak area	Concentration found (µg/mL)	Average	SD	RSD%	
80	1942307	79.477	79.475	0.149	0.187	
80	1945764	79.623	-	-	-	
80	1938729	79.325	-	-	-	
100	2475997	102.092	101.585	0.440	0.433	
100	2458935	101.369	-	-	-	
100	2457207	101.296	-	-	-	
120	2894210	119.813	119.862	0.073	0.061	
120	2894544	119.828	-	-	-	
120	2897362	119.947	-	-	-	

SD: Standard deviation (n: 3), RSD: Relative standard deviation

Table 2b. Inter-day precision						
Concentration (µg/mL)	Peak area	Concentration found (µg/mL)	Average	SD	RSD%	
80	1939789	79.370	79.417	0.043	0.054	
80	1941781	79.455	-	-	-	
80	1941161	79.428	-	-	-	
100	2442327	100.665	101.802	1.005	0.986	
100	2487270	102.569	-	-	-	
100	2477898	102.172	-	-	-	
120	2894277	119.816	119.868	0.157	0.131	
120	2892563	119.743	-	-	-	
120	2899665	120.044	-	-	-	

SD: Standard deviation (n: 3), RSD: Relative standard deviation

Table 2c. Repeatability					
Concentration (µg/mL)	Peak area	Concentration found (µg/mL)	Average	SD	RSD%
100	2443258	100.704	100.959	0.250	0.248
100	2448178	100.913	-	-	
100	2446070	100.824	-	-	
100	2444889	100.774	-	-	-
100	2457432	101.305	-	-	-
100	2455724	101.232	-	-	-

System suitability

The system suitability test is an important element of chromatographic analysis since it ensures that the chromatographic system's accuracy and repeatability are sufficient for analysis. It was performed with six replicate injections of the standard solution of olaparib. The retaining of olaparib was found to be 4.32 min, having a tailing factor of not more than 1.17 in all peaks, indicating good peak symmetry. Theoretical plates were found to be 3160. The results are reported in Table 7.

Detection limit and quantification limit

The LOD and LOQ of olaparib were found to be 0.49 μ g/mL and 1.49 μ g/mL, respectively.

Assay of standard formulation of lynparza (olaparib)

Assay validation provides reliability assurance during normal use, and is sometimes referred to as "the process of providing documented evidence that the method is doing what it intends to do". The purity by HPLC is determined by the percentage recovery of olaparib. The developed method was very accurate, precise and robust as recovery percentage was within 100 ± 2 given in Figure 3a and Table 8.

Analytical solution stability

The olaparib sample solution was stable for 24 h at room temperature. The stability results were analyzed for the percentage difference from zero time injection, where there was no decrease in the peak areas of the drug nor a shifting of retention time was detected. The observations obtained from the stability phenomenon are reported in Table 9.

Force degradation study

The drug degradation study was conducted in acid, alkaline, and oxidation solutions to determine the stability of the drug under different conditions.

Acid degradation

The acid degradation of olaparib was carried out at different concentrations of (1-5 M) HCl until it was degraded for a period of 30 min. The degraded chromatogram and the non-degraded chromatogram were compared and the percentage of degradation was calculated. The chromatogram is given in (Figure 3b) and the results are summarized in Table 10. Acid degradation was found to be 12.69% for 5 M HCl.

Table 3. Accuracy observation table						
Recovery level	API added (mg)	API recovered (mg)	Recovery%	Average recovery%	RSD%	
90%	-	22.110	98.680	-	-	
90%	24.900	22.100	98.610	98.650	0.230	
90%	-	22.500	100.390	-	-	
100%	-	24.720	99.290	-	-	
100%	24.900	24.920	100.070	99.680	0.140	
100%	-	25.060	100.000	-	-	
110%	-	26.920	98.300	-	-	
110%	24.900	26.900	98.220	98.260	0.050	
110%	-	26.870	98.110	-	-	

RSD: Relative standard deviation, API: Active pharmaceutical ingredient

Table 4. Sp	ecificity table	e of olaparib					
Analyte	Added%	Excipient amount added (mg)	Concentration found (µg/mL)	Recovery%	Average recovery%	SD	RSD%
Olaparib	50	5	100.705	100.705	100.814	0.105	0.104
(10 mg)	50	5	100.913	100.913	-	-	-
	50	5	100.824	100.824	-	-	-
	100	10	100.774	100.774	100.915	0.342	0.340
	100	10	101.305	101.305	-	-	-
	100	10	100.665	100.665	-	-	-
	150	15	102.569	102.57	101.710	0.892	0.877
	150	15	100.789	100.789	-	-	-
	150	15	101.773	101.774	-	-	-

Alkali degradation

The alkali degradation of olaparib was carried out at different concentrations of (1-5 M) NaOH until it was degraded. The degraded chromatogram and the non-degraded chromatogram were compared and the percentage of degradation was calculated. The chromatogram is given in Figure 4a and the results are summarized in Table 9. Alkali degradation was found to be 2.60% at 5 M NaOH.

Peroxide degradation

Peroxide degradation of olaparib was carried out at a concentration of 30%. A comparative study of the peroxide degraded olaparib chromatogram and the non-degraded chromatogram was conducted to calculate the 2.55% degradation. The chromatogram is given in Figure 4b and the results are summarized in Table 10.

Thermal degradation

Thermal degradation of the drug was found to be negligible. The olaparib drug was found to be thermal stable as there was no degradation when exposed to thermal conditions. The chromatogram is given in Figure 5a, and the results are summarized in Table 10.

Photolytic degradation

Degradation by photolysis of olaparib was found to be negligible. The olaparib drug was found to be light stable as there was no degradation when exposed to light. The chromatogram is given in (Figure 5b) and the results are summarized in Table 10.

DISCUSSION

Olaparib is a new drug, so almost no method is available to estimate olaparib in bulk and pharmaceutical dosage form. Therefore, our present aim was to develop a new, compatible, stable, robust method for the determination of olaparib in bulk and formulations by RP-HPLC.

According to the ICH guidelines, the developed method was validated for the following parameters: system suitability, linearity, accuracy, precision, robustness, and analytical

Table 5.	Table 5. Robustness study with flow rate, pH, and mobile phase composition						
Sample ID	Analytical condition	Olaparib input (mg)	Olaparib recovery (mg)	Olaparib recovery (%)	Mean recovery olaparib (%)	SD	RSD%
_	Flow rate: 1.1 mL/min	_			100.060	0.680	0.680
1	Mobile phase pH: 3.5	— 2F	24.040	00.500			
'	Mobile phase ratio: 50:50	[—] 25 —	24.860	99.500			
	Column: C18 (150 mm x 4.6 mm, 5 µm)				_		
	Flow rate: 0.9 mL/min	_					
2	Mobile phase pH: 3.5	- 25	25.090	100 200			
_	Mobile phase ratio: 50:50		25.090	100.300			
	Column: C18 (150 mm x 4.6 mm, 5 μm)				_		
	Flow rate: 1 mL/min	_		99.200			
3	Mobile phase pH: 3.6	_	24.820				
3	Mobile phase ratio: 50:50	— 25 —					
	Column: C18 (150 mm x 4.6 mm, 5 μm)	18 (150 mm x 4.6 mm, 5 μm)		_			
	Flow rate: 1 mL/min	_					
4	Mobile phase pH: 3.4	— 25	25.110	100.400			
7	Mobile phase ratio: 50:50						
	Column: C18 (150 mm x 4.6 mm, 5 μm)				_		
	Flow rate: 1 mL/min	_					
5	Mobile phase pH: 3.5	[—] 25	24.980	00.000			
	Mobile phase ratio: 55:45	_	24.900	99.900			
	Column: C18 (150 mm x 4.6 mm, 5 μm)				_		
	Flow rate: 1 mL/min	_					
6	Mobile phase pH: 3.5	— 2F	25.290	101100			
-	Mobile phase ratio: 45:55	[—] 25 —	∠J.∠7U	101.100			
	Column: C18 (150 mm x 4.6 mm, 5 µm)						

solution stability. The RSD% value was well below 2 and the percentage recovery was within the limit of 100 ± 2 . The stability of the drugs is a immense issue during formulation and still no

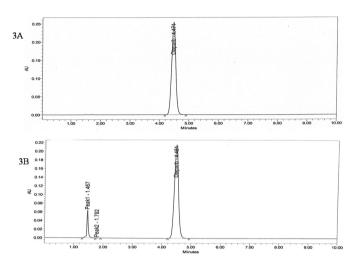


Figure 3. A) Chromatogram showing assay of standard formulation, B) Chromatogram of acid-degraded olaparib

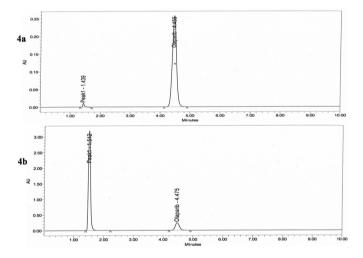


Figure 4. a) Chromatogram of base degraded olaparib, b) Chromatogram of hydrogen peroxide degraded olaparib

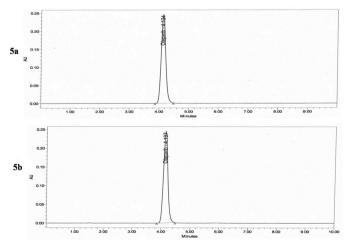


Figure 5. a) Chromatogram of thermal degraded olaparib, b) Chromatogram of photolytic degraded olaparib

stability data have been reported. To determine the stability of olaparib, we conducted a force degradation study.

The drug was found to be very stable, when exposed to heat and light. It was also found to be quite stable under both acidic and basic conditions. Higher concentrations of acids and bases (5 M) could degrade the drug, too, by 12.69% under acidic conditions, whereas in the case of basic and oxidation

Table 6. Ruggedness of olaparib						
S. no.	Analyst 1 (assay%)	Assay percent olaparib	Analyst 2 (assay%)			
1	99.390		100.560			
2	100.210		99.640			
3	98.040		99.800			
4	102.690		100.410			
5	99.980		100.200			
6	101.560		97.100			
Mean	100.480		99.620			
RSD% Overall percentage RSD	1.660	1.480	1.290			

RSD: Relative standard deviation

Table 7. System suitability parameters and achieved values parameters

values achieved in the validation phenomenon				
Theoretical plates	3160			
Retention time	4.32 minutes			
Asymmetry	1.170			
RSD	0.240%			

RSD: Relative standard deviation

Table 8. Summery of validation		
Parameter	Acceptable limit	Remark
Accuracy	98.260-99.680	Passed
Specificity	0.100-0.800	Passed
Precision	0.056-0.980	Passed
Linearity	0.998	Passed
Ruggedness and Robustness	1.480 and 0.680	Passed
Assay	98.680%	Passed

Table 9. Solution stability data						
Time	Inject	Time	Difference (sec)	Area	Calculation	
Initial	9	2	0	2442536	0.00	
After 1 h	10	9	67	2450664	0.00	
After 24 h	10	5	1436	2473824	-0.01	

Table 10. Force degradation data			
Type of degradation	Area	Degradation%	Peak purity
No degradation (standard chromatogram 100 μg/mL)	2443258	-	Passed
2.5 mL of 5 M HCl (acidic)	2133134.666	12.690	Passed
2.5 mL 5 M NaOH (alkaline)	2507018.666	2.600	Passed
1 mL of 30% $\rm H_2O_2$ (peroxide)	2380875	2.550	Passed
Thermal degradation	2443258	0	Passed
Photolytic degradation	2443258	0	Passed

conditions, degradation was found to be 2.60% and 2.55%. Under thermal and photolytic conditions, they were found to be stable.

CONCLUSION

The developed method in RP-HPLC was established to be simple, cost-effective, accurate, and robust; so that, it can be extensively applied for estimating any formulations of olaparib. This degradation study was conducted on HPLC for the first time. As the HPLC method is less cost-effective compared with the LC-MS method, this will prove an effective method for estimating olaparib.

Degradation studies were conducted and from the study we can conclude that the degradation of olaparib was very low in the case of basic and oxidation conditions, whereas in the case of acidic degradation by HCl, the highest degradation was observed. The degradation was found to be 12.69%. No degradations were found in the case of thermal and photolytic conditions. The drug olaparib can be considered a very stable drug in all conditions except the acidic condition.

ACKNOWLEDGMENTS

The authors are grateful to the Department of Delhi Pharmaceutical Sciences and Research University, New Delhi (India) for providing the facilities for working and to the laboratory assistance.

Ethics

Ethics Committee Approval: Not applicable.

Informed Consent: Not applicable.

Peer-review: Externally peer-reviewed.

Authorship Contributions

Concept: A.C., R.T., Design: A.C., R.T., S.D., Data Collection or Processing: A.C., P.D., Analysis or Interpretation: M.G., S.D., A.C., P.D., Literature Search: A.C., P.D., Writing: M.G., S.D., A.C., P.D.

Conflict of Interest: No conflict of interest was declared by the authors.

Financial Disclosure: The authors declared that this study received no financial support.

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